

“VALIDATION AND IMPLEMENTATION OF AN ANALYTICAL METHODOLOGY TO CHARACTERIZE THE FRACTION OF NANO-SIZED PARTICLES IN E171 AND E174”

(Nanofood@, nanoAg@ and EFSA nano projects)

Jan Mast

Service Trace Elements and Nanomaterials
SD Chemical and physical health risks, Sciensano

EFSA Nano Stakeholders Engagement Event, 1 & 2 April 2019, Parma

.be

OUTLINE

- **Introduction**
- **Analytical methodology**
- **Analysis results of E174**
- **Analysis results of E171**
- **Lessons learnt**

Characterization of NM in complex matrices in a regulatory context

Food

Environment

Medical devices

Cosmetics

Characterization of NM in food additives E171, E174, E175

Nanofood@

NanoAg@

EFSA

EM unit

Objectives

Validated methods

Market study

Exposure assessment

“EFSA Guidance on risk assessment of the application of nanoscience and nanotechnologies in the food and feed chain”

“4. Physicochemical characterization of nanomaterial”

- “4.1. Framework for distinguishing nanomaterials and non-nanomaterials”
- “4.2 Pristine material characterization”
- “4.3. Characterization and quantification in matrix” (products)

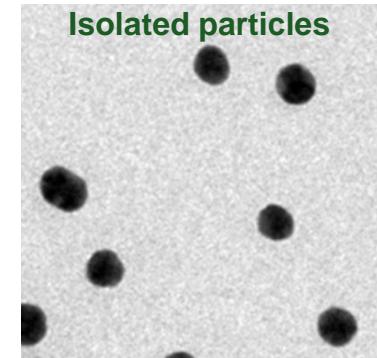
« Through validated methods, suitable analytical techniques »

- BF-TEM
- SEM-EDX + STEM-EDX
- SP-ICP-MS

Research question: «Materials falls under scope of EFSA guidance (section 1.3)?»

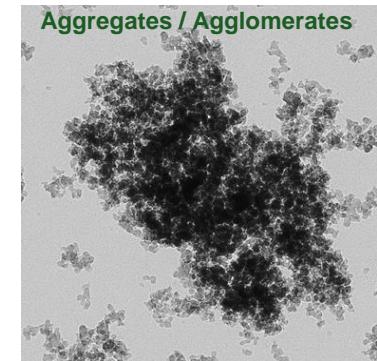
EC recommendation on the definition of a nanomaterial (2011/696/EU):

- Natural, incidental, and manufactured
- Refers to constituent particles (primary particles)
- Uses size as the most important defining parameter: “50 % or more of the constituent particles (by number) have one or more external dimensions in the size range 1 – 100 nm.”



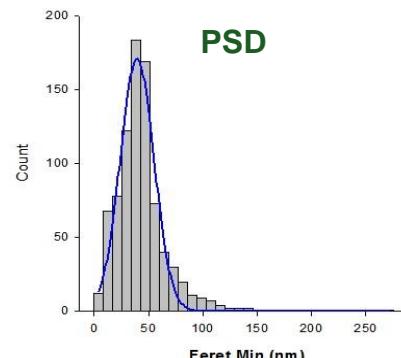
Novel food Regulation (2015/2283):

- Engineered material, intentionally produced to perform a specific function in the food.
- One or more dimensions of the order of 100 nm or less
- It includes also structures, agglomerates, or aggregates, which may have a size above 100 nm but retain properties that are characteristic of the nanoscale.



+ EU Regulation on the Provision of Food Information to Consumers

“all ingredients that are present in food as nanomaterials must be indicated as such on the product label.” (refers to Novel food regulation)



OUTLINE

- **Introduction**
- **Analytical methodology**
- **Analysis results of E174**
- **Analysis results of E171**
- **Lessons learnt**

Identification of nano-objects in food

CEN/TC 352: Nanotechnologies — Guidance on detection and identification of nano-objects in complex matrices

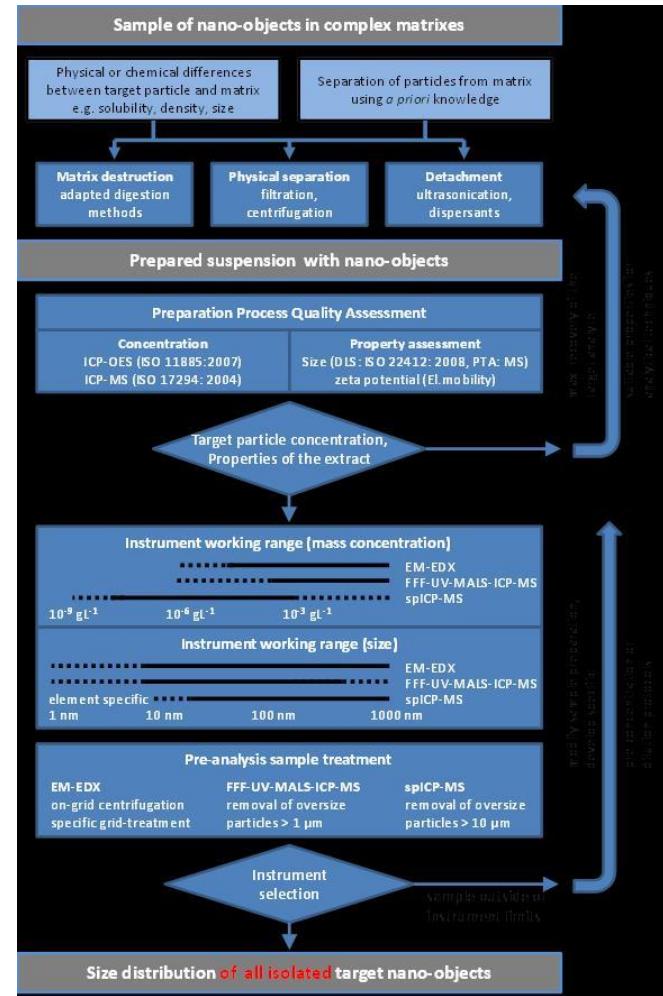
Liquid environmental compartments

- waste water
- consumer products
- Food
- cosmetics

Sample preparation

Size classification +chemical composition

- EM: BF-TEM, SEM-EDX or STEM-EDX
- SP-ICP-MS
- FFF + multiple detectors



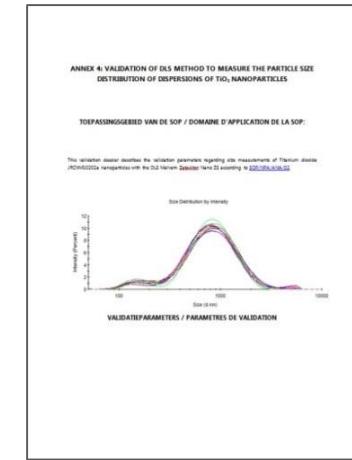
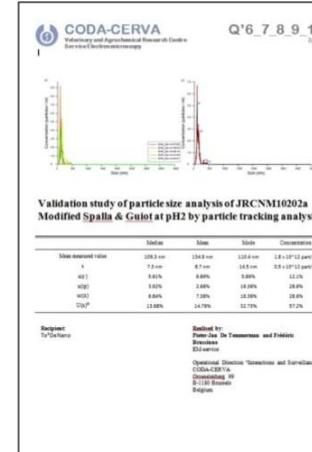
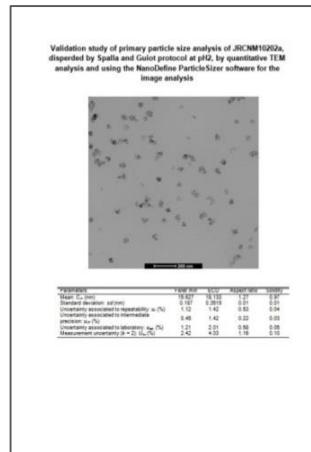
Analytical approach

Sub-question	Methodology	Illustration
Particles present?	Descriptive (conventional) EM	
Phase?	Powder electron diffraction	
Chemical composition?	HAADF-STEM-EDX SEM-EDX	
Nano?	Quantitative TEM	
How many?	Total: ICP-MS NP: SP-ICP-MS	

Quality of the results

Standardised methodology

Validated methodologies



Detailed physicochemical characterization

Technical data sheet of the representative test material JRCNM10202a¹

Frédéric Brasseur, Pieter-Jan De Temmerman, Jan Mest (CODA-CERVA)

Chemical identification of the substance

Chemical name of the substance	Titanium oxide
Chemical formula	TiO ₂
CAS Number	13463-67-7
EC number	236-673-5
REACH registration number	Not available

Particle shape and size²

Determination methods	TEM ³
Shape	Sphere
Number of dimensions	2D
Characteristic dimension	Diameter (Feret Min)
Mean particle size	17 nm
Median particle size	16 nm
Standard deviation	1.2 nm
Measurement uncertainty	1.7% (95% CI) ⁴
Agglomerates and Aggregates ⁵	
Is the substance available in aggregated form?	Yes
Determination of the aggregate size	Together with the aggregated particles
Is the substance available in aggregated form?	Yes

Determination method

Size measured as	Mean size and standard deviation (Feret Min)
Conditions ⁶	Stable (pH 7.5)
Mean size	33 nm
Standard deviation	1.5 nm
Shape description	Rounded
Measurement uncertainty	24% (95% CI) ¹

Determination method

Size measured as	Mean hydrodynamic radius and standard deviation
Conditions ⁶	Stable (pH 7.5)
Mean size	109 nm
Standard deviation	7 nm

Determination method

Size measured as	Mean hydrodynamic radius
Conditions ⁶	Stable (pH 7.5)
Mean size	632 nm
Measurement uncertainty	2% (95% CI) ¹

Technical data sheet of the representative test material JRCNM102000a¹

Frédéric Brasseur, Pieter-Jan De Temmerman, Jan Mest (CODA-CERVA)

Chemical identification of the substance

Chemical name of the substance	Titanium oxide
Chemical formula	TiO ₂
CAS Number	13463-86-9
EC number	231-645-4
REACH registration number	Not available

Particle shape and size²

Determination methods	TEM ³
Shape	Sphere
Number of dimensions	3
Characteristic dimension	Diameter (Feret Min)
Mean particle size	117 nm
Median particle size	112 nm
Standard deviation	4 nm
Measurement uncertainty	5% (95% CI) ¹

Agglomerates and Aggregates⁵

Is the substance available in aggregated form?	Yes
Determination of the aggregate size	Together with the aggregated particles
Is the substance available in aggregated form?	Yes

Determination method

Size measured as	Mean size and standard deviation (Feret Min)
Conditions ⁶	Stable (pH 7.5)
Mean size	117 nm
Standard deviation	10 nm

Determination method

Size measured as	Mean hydrodynamic radius
Conditions ⁶	Stable (pH 7.5)
Mean size	112 nm
Measurement uncertainty	24% (95% CI) ¹

Determination method

Size measured as	Mean hydrodynamic radius
Conditions ⁶	Stable (pH 7.5)
Mean size	950 nm
Measurement uncertainty	1% (95% CI) ¹

OUTLINE

- **Introduction**
- **Analytical methodology**
- **Analysis results of E174**
- **Analysis results of E171**
- **Lessons learnt**

Silver (E174) in food

- EC approved food additive (EC 1333/2008)
- Authorized to be used
 - to color the external coating of confectionery
 - for decoration of chocolates
 - in liqueurs
- Applied as a silver-colored powder or as tiny sheets (Directive 95/45/EC)
- Intended and assumed to be present in the bulk form
- EFSA re-evaluation of silver (E 174) as food additive
- Conclusions warrant further characterization of nano-sized fraction using EM



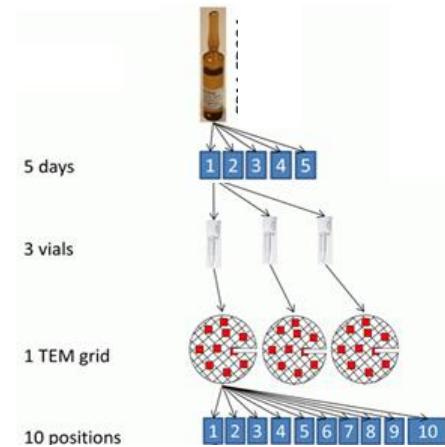
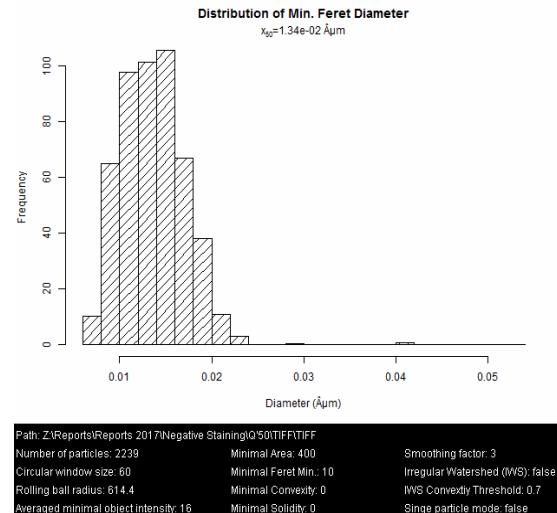
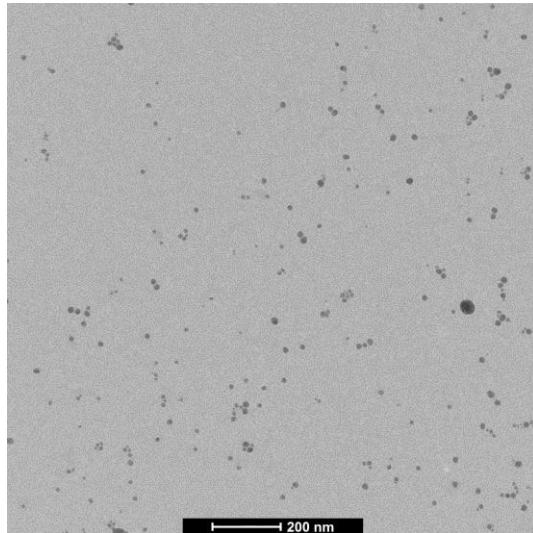
Purchase of E174 materials

Belgian market

- Colloidal reference materials
- Pristine materials purchased from different European websites
 - ➔ Powders, flakes, petals, leaves
- E174 containing food products purchased from several local stores
 - ➔ Sugar beans, candy, cake decoration



Validation based on reference material: NM-300k



Uncertainty budget of quantitative TEM analysis

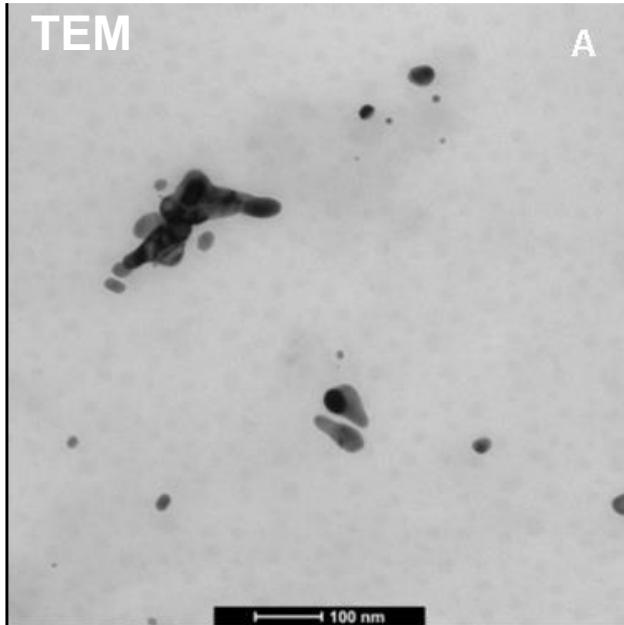
- Type A uncertainties: top-down estimation of U_r , U_{ip} (5 days, 3 repetitions)
- Type B uncertainties: U_{cal} , U_t
- Combined, expanded measurement uncertainty ($k = 2$): $U_{cx} = 8\text{-}9\%$

E174 in food products

Mini confetti, silver

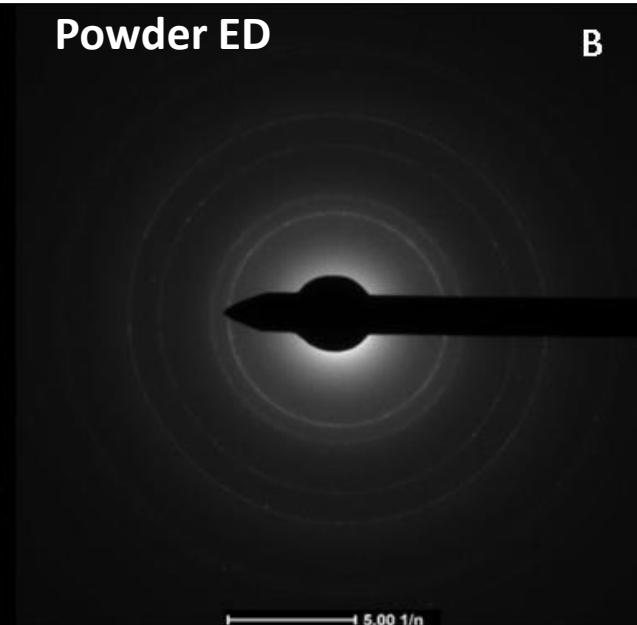


TEM



A

Powder ED



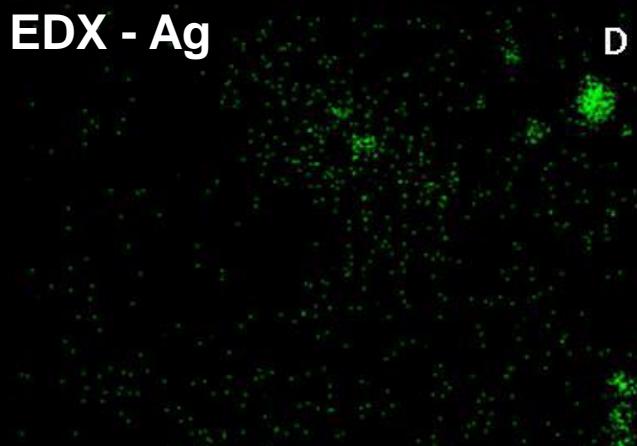
B

SEM



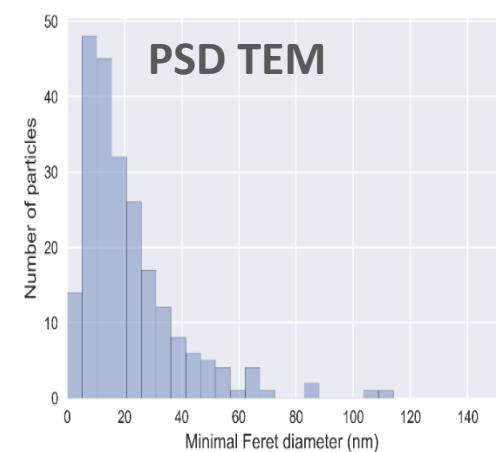
C

EDX - Ag

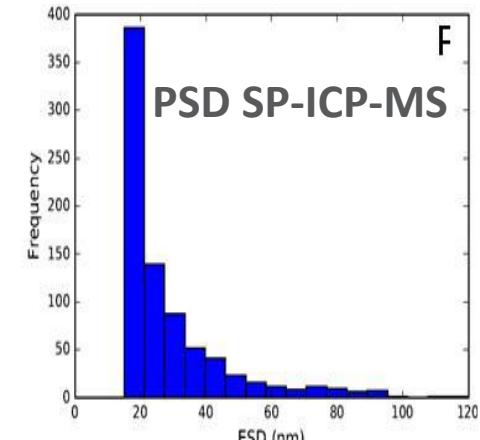


D

PSD TEM



PSD TEM



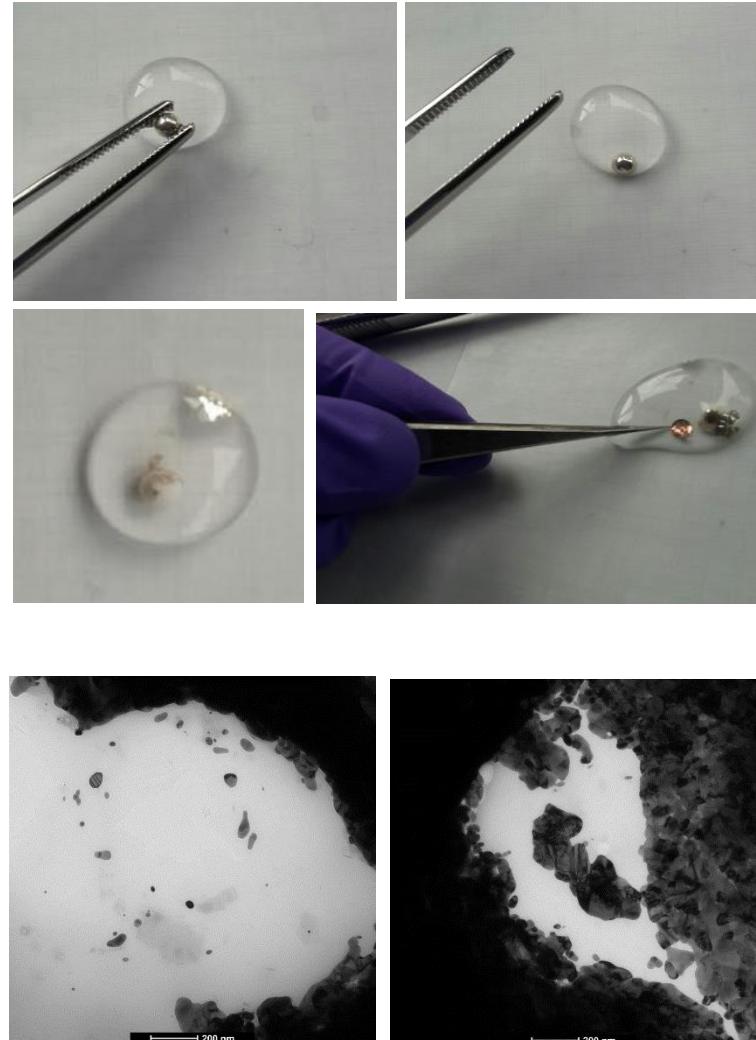
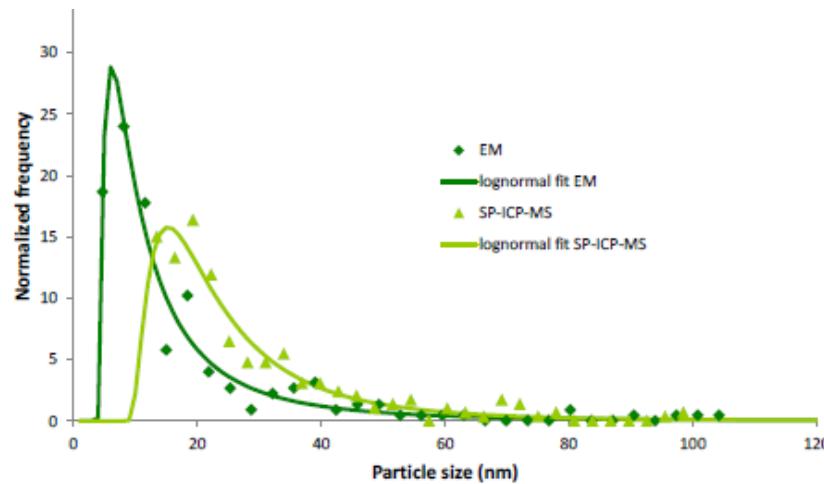
PSD SP-ICP-MS

E174 food products

Reference	Feret Min (TEM)*		ESD (SP-ICP-MS) *	# NP (<120nm) (/kg)	Mass NP (<120nm) (mg/kg)	Total Ag mass (g/kg)	% Ag NP mass of total Ag mass	#NP/ kg Ag
	Non-sonicated	sonicated	sonicated					
NF@-Mix-Ag-001	19 nm	-	24 nm	7.1x10 ¹³	21	4.7	0.46	1,51x10 ¹⁶
NF@-Mix-Ag-002	17 nm	-	19 nm	3.5x10 ¹³	6.9	4.7	0.15	7,45x10 ¹⁵
NF@-Mix-Ag-003	16 nm	-	22 nm	4.2x10 ¹³	11	3.4	0.32	1,24x10 ¹⁶
NF@-Mix-Ag-004	20 nm	-	22 nm	4.5x10 ¹³	13	4.1	0.32	1,10x10 ¹⁶
NF@-Mix-Ag-005	15 nm	-	23 nm	7.4x10 ¹³	23	4.7	0.49	1,57x10 ¹⁶
NF@-Mix-Ag-006	36 nm	-	16 nm	1.6x10 ¹⁴	32	8.7	0.37	1,84x10 ¹⁶
NF@-Mix-Ag-007	22 nm	-	22 nm	3.6x10 ¹³	9.7	4.5	0.22	8,00x10 ¹⁵
NF@-Mix-Ag-008	23 nm	-	21 nm	1.7x10 ¹³	3.4	1.2	0.28	1,42x10 ¹⁶
NF@-Mix-Ag-010	12 nm	-	20 nm	7.8x10 ¹³	22	4.2	0.52	1,86x10 ¹⁶
NF@-Mix-Ag-011	10 nm	-	22 nm	2.0x10 ¹³	6.3	1.9	0.33	1,05x10 ¹⁶

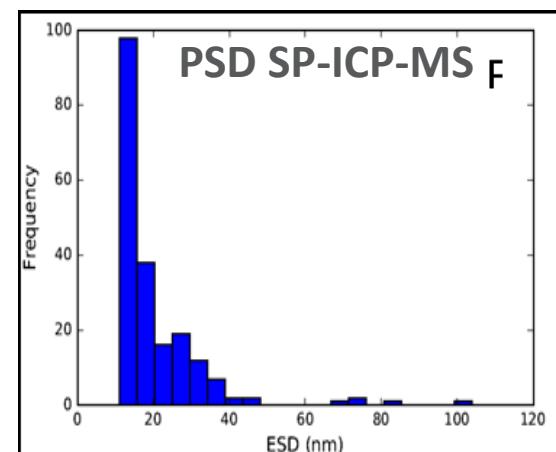
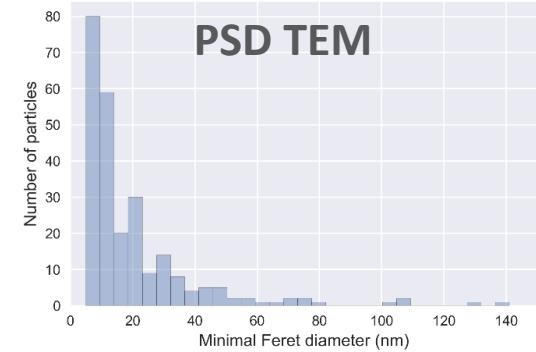
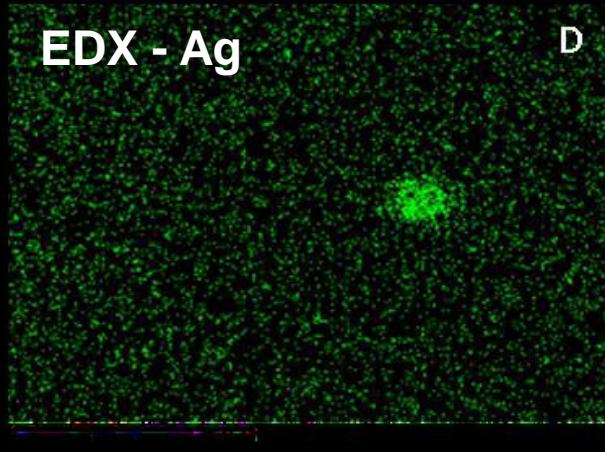
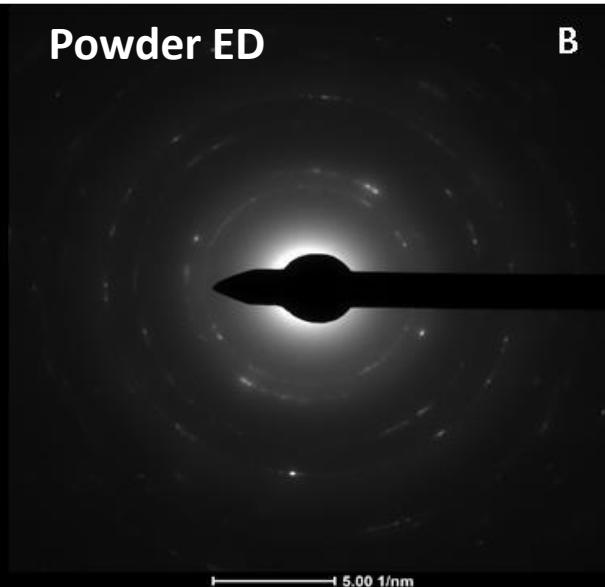
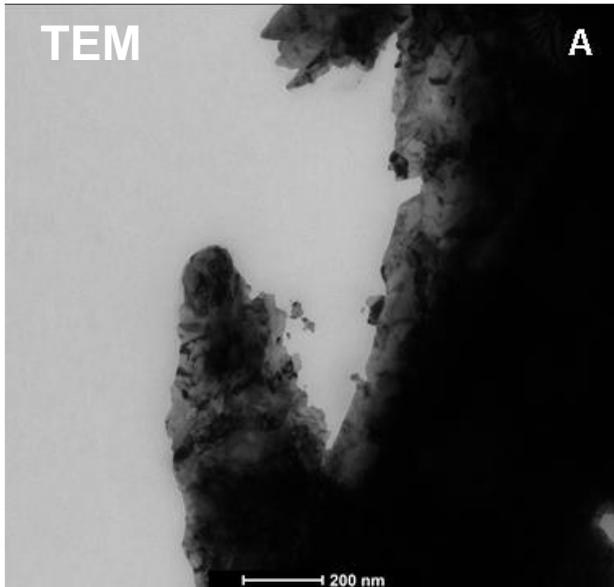
Results E174 in food products

- Sample preparation relatively easy
- NPs readily observed by EM
- Flakes with rough, particle-like surface
- Even distribution over EM grid → quantitative TEM
- Representative size distributions, Median values of Feret min values: ± 10-20 nm
- (SP-ICP-MS: 10^{15} - 10^{16} NP per kg E174)



Pristine E174

"Silber-gastro Flocken"



Pristine E174 materials

Reference	Feret Min (TEM)		ESD (SP-ICP-MS)	NP (<120nm) mass (g/kg)	% Ag NP mass of total Ag mass	#NP (<120nm) /kg Ag
	Non-sonicated	sonicated	sonicated			
NF@-Ag-001	13 nm	13 nm	14 nm	385	38.5	2.0x10 ¹⁹
NF@-Ag-002	12 nm	Insufficient number	Insufficient number	Insufficient number	-	<1.4x10 ¹⁵
NF@-Ag-003	±20 nm	Insufficient number	16 nm	2.3	0.23	1.5x10 ¹⁶
NF@-Ag-004	Insufficient number	10 nm	14 nm	2.5	0.25	2.8x10 ¹⁶
NF@-Ag-005	±20 nm	Insufficient number	27 nm	2.7	0.27	7.1x10 ¹⁵
NF@-Ag-006	±45 nm	Insufficient number	16 nm	5.0	0.5	3.8x10 ¹⁶
NF@-Ag-007	Insufficient number	12 nm	Insufficient number	Insufficient number	-	<9.5x10 ¹⁵
NF@-Ag-008	Insufficient number	14 nm	20 nm	2.2	0.22	1.3x10 ¹⁶
NF@-Ag-009	7 nm	17 nm	17 nm	2.2	0.22	1.2x10 ¹⁶
NF@-Ag-010	14 nm	11 nm	21 nm	4.5	0.45	2.1x10 ¹⁶
NF@-Ag-013	5 nm	14 nm	17 nm	6.2	0.62	3.5x10 ¹⁶
NF@-Ag-014	12 nm	13 nm	15 nm	3.2	0.32	2.5x10 ¹⁶

Results pristine E174

Difficult to obtain representative size distributions

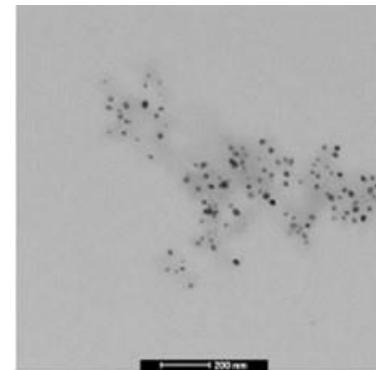
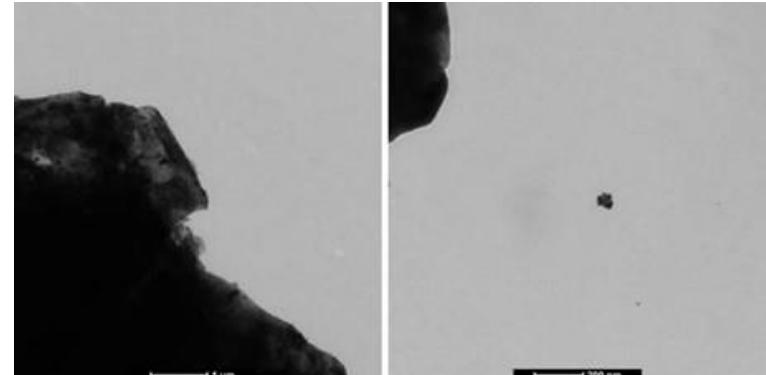
- Few particles
- Non-homogeneous particle distribution
- Flakes are large and smooth
- Some samples contain higher amount of NPs and colloidal fraction

SP-ICP-MS: 10^{15} - 10^{16} NP per kg E174

Median values of Feret min values: ± 10 - 20 nm

Reactivity of silver = bottleneck sample preparation

- ⇒ Tailored sample preparation methods for dispersing Ag may lead to artefacts (Demonstrated for pH, PVP, acetone, sonication)
- ⇒ Examined Ag materials react differently



Conclusions E174

Few nanosized particles in pristine E174

Larger nano-sized fraction in products containing E174

Hypothesis: processing of E174 in products induces formation silver NP

- More nano-sized particles in the products than in the pristine materials
- The surface structure of E174 in products is altered (release associated?)
- Mechanisms of formation of NPs :
 - *De novo* formation from Ag^+ by chemical reactions between E174 and reducing agents (PVP, acetone, proteins,...) in the dispersion medium
 - Physical treatment, mimicked by (too high levels of) sonication energy,

For risk analysis:

- Applied amount of pristine E174
- Modification of E174 (particle) properties induced by processing

OUTLINE

- **Introduction**
- **Analytical methodology**
- **Analysis results of E174**
- **Analysis results of E171**
- **Lessons learnt**

Titanium dioxide (E171) in food

- EC approved food additive (EC 1333/2008)
- Authorized to be used as color in foodstuffs, widely used
- Applications applied on refractive properties (shiny coating, UV protection)
- Several crystal structures, such as anatase and rutile
- Intended and assumed to be present in the bulk form



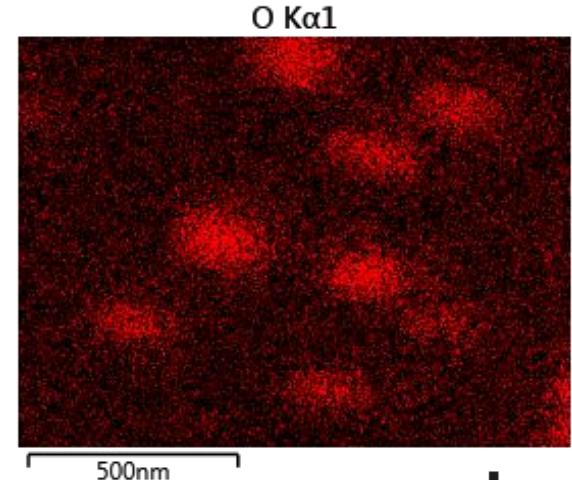
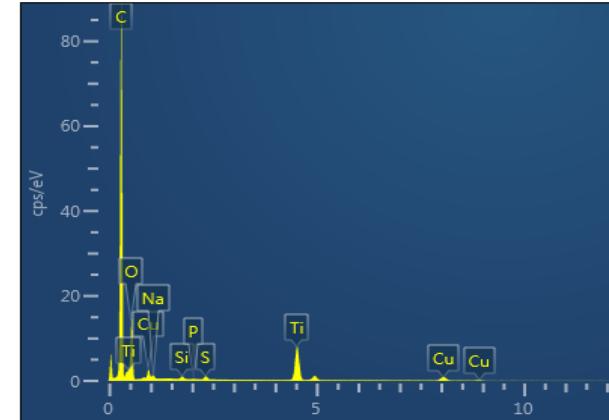
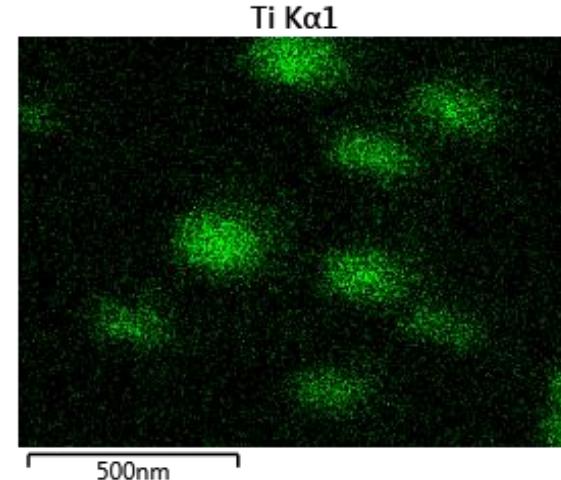
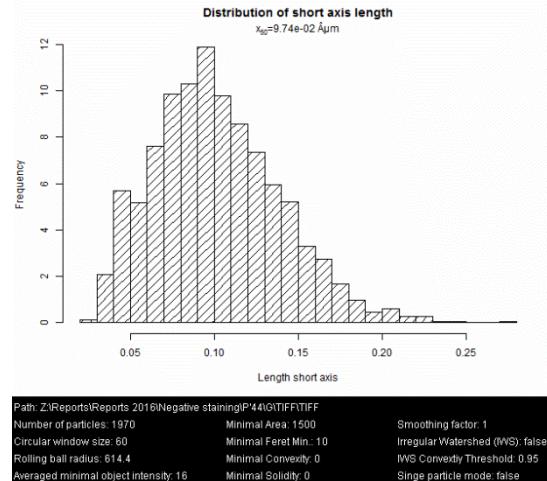
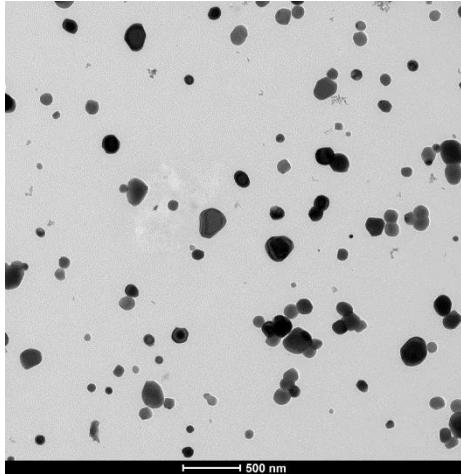
Purchase of E171 materials

Belgian market

- Reference materials
- Pristine materials purchased from different European websites
 - ➔ Powders, flakes
- E171 containing food products purchased from several local stores
 - ➔ sugarbeans, candy, chewing gum, cake decoration, ...

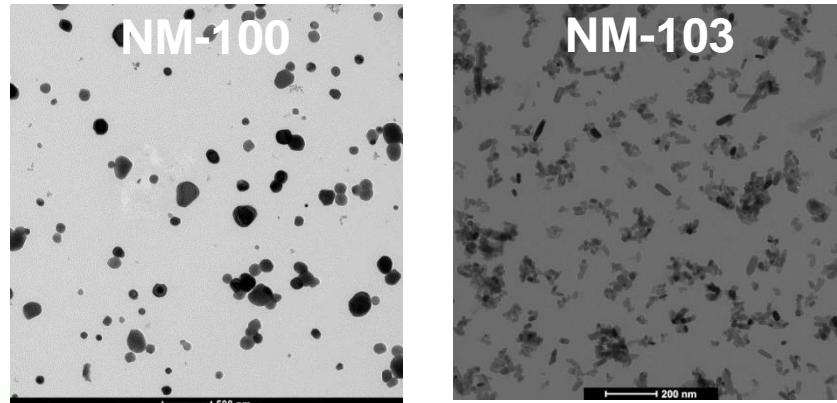
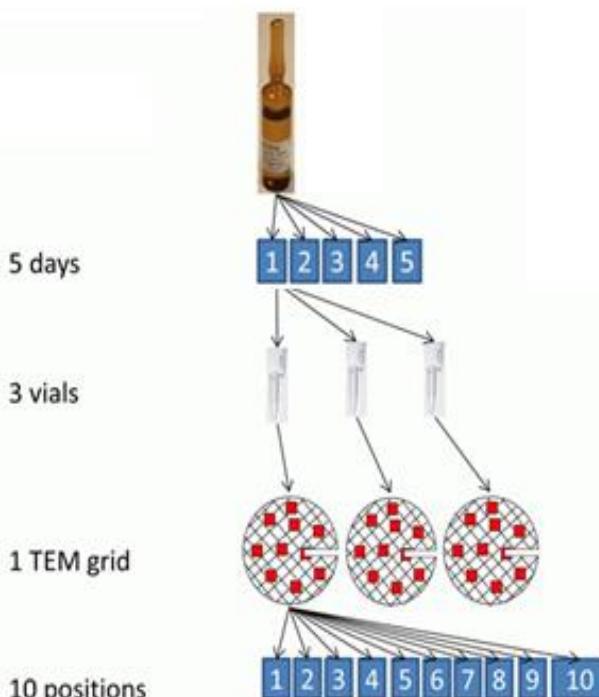


Reference materials: NF@-Ti-002, and RTMs NM-100 and NM-103



Size: TEM: median Feret min: NF@-Ti-002: 94.8 nm, NM-100: 100.1 nm, NM-103: 18.2 nm
SP-ICP-MS: median ESD: NF@-Ti-002: 90 nm, NM-100: 107 nm

Measurement uncertainties based on reference materials

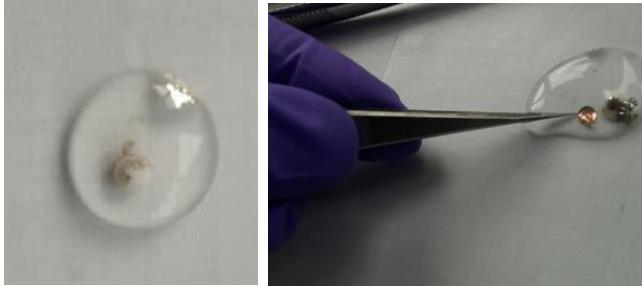
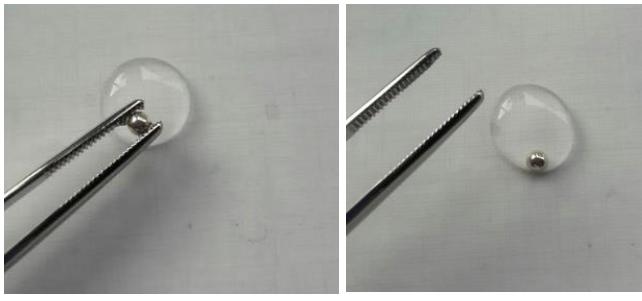
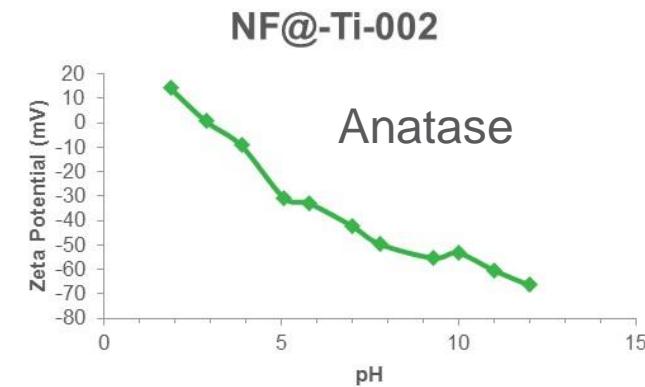
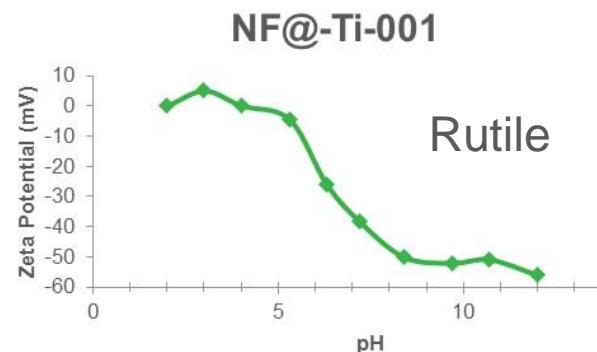
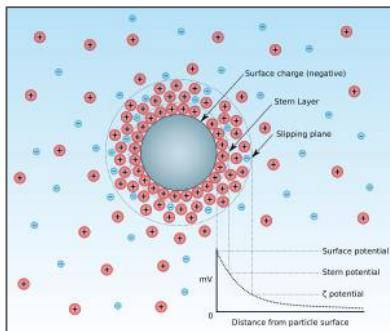


Parameter: median Feret Min	NM-100	NM-103
Mean: C_m (nm)	100.1	18.2
Standard deviation: sd (nm)	2.0	0.4
Uncertainty associated to repeatability: u_r (%)	1.9	2.1
Uncertainty due to day-to-day variation: u_{day} (%)	0.7	1.3
Uncertainty associated to intermediate precision: u_{IP} (%)	2.1	2.5
Uncertainty associated to calibration: u_{cal} (%)	0.8	0.1
Uncertainty associated to trueness: u_{tr} (%)	3.6	3.9
Combined uncertainty: $u_c(x)$ (%)	4.3	4.6
Measurement uncertainty ($k = 2$): U_{cx} (%)	8.5	9.2

Top-down estimation of the measurement uncertainties:
TEM: measurement uncertainty ($k = 2$): $U_{cx} = 8-10\%$ 
SP-ICP-MS: measurement uncertainty ($k = 2$): $U_{cx} = 16\%$

Sample preparation

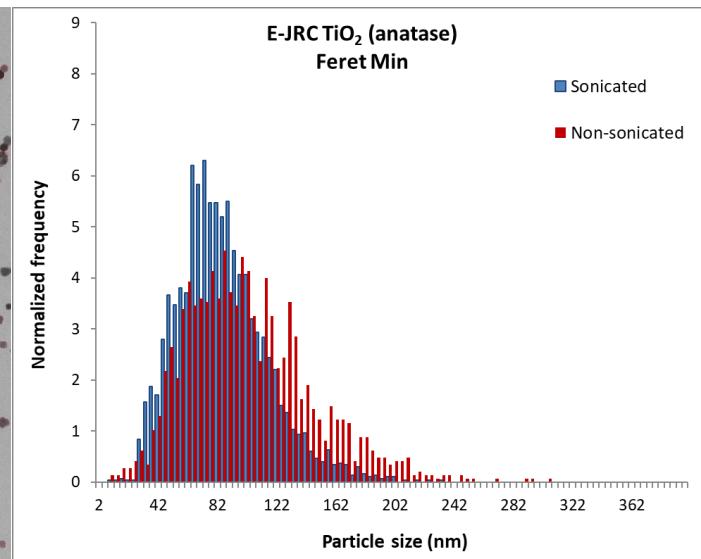
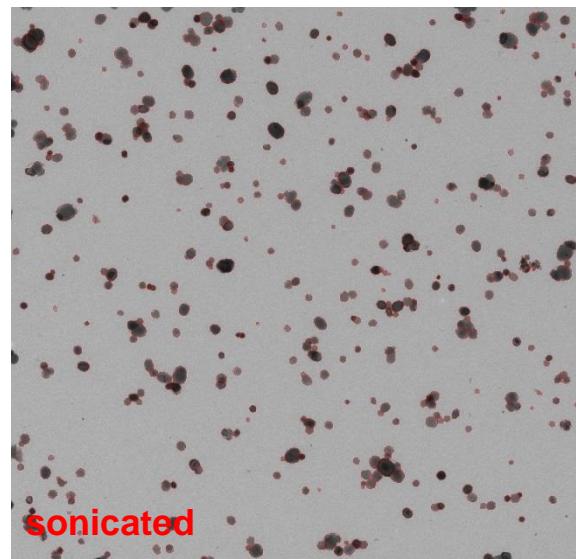
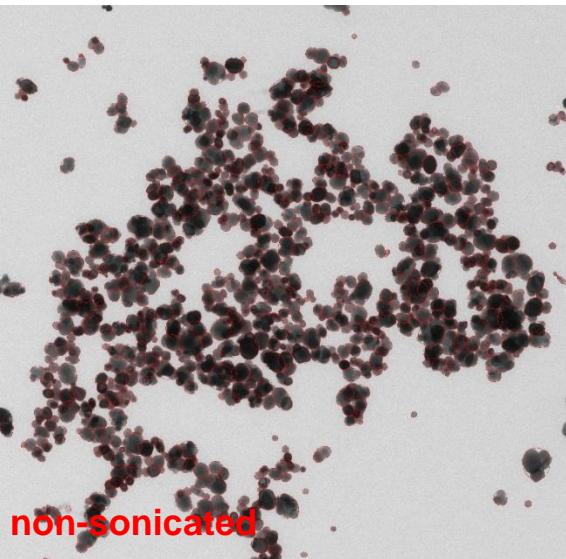
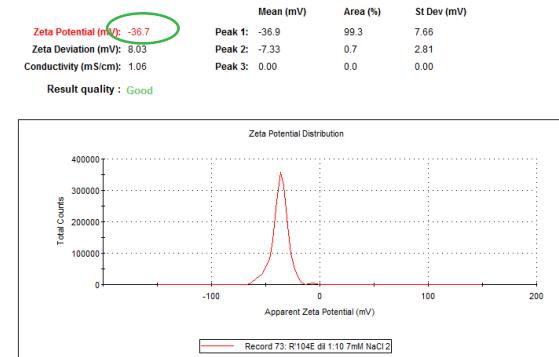
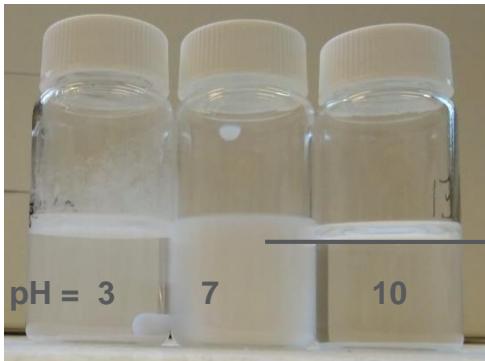
De-agglomeration by pH adjustment



Calibrated probe
-sonication



Illustration: pristine TiO₂, anatase



of 9 March 2012

laying down specifications for food additives listed in Annexes II and III to Regulation (EC) No 1333/2008 of the European Parliament and of the Council

(Text with EEA relevance)

(OJ L 83, 22.3.2012, p. 1)

E 171 TITANIUM DIOXIDE

Synonyms

CI Pigment White 6

Definition

Titanium dioxide consists essentially of pure anatase and/or rutile titanium dioxide which may be coated with small amounts of alumina and/or silica to improve the technological properties of the product.

The anatase grades of pigmentary titanium dioxide can only be made by the sulphate process which creates a large amount of sulphuric acid as a by-product. The rutile grades of titanium dioxide are typically made by the chloride process.

Certain rutile grades of titanium dioxide are produced using mica (also known as potassium aluminum silicate) as a template to form the basic platelet structure. The surface of the mica is coated with titanium dioxide using a specialised patented process.

Rutile titanium dioxide, platelet form is manufactured by subjecting titanium dioxide (rutile) coated mica nacreous pigment to an extractive dissolution in acid followed by an extractive dissolution in alkali. All of the mica is removed during this process and the resulting product is a platelet form of rutile titanium dioxide.

laying down specifications for food additives listed in Annexes II and III to Regulation (EC) No 1333/2008 of the European Parliament and of the Council

(Text with EEA relevance)

Anatase

2, p. 1)

white 6

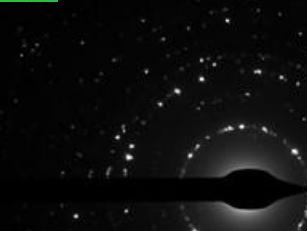
ide consists essentially of pure anatase and/or rutile
ide which may be coated with small amounts of
r silica to improve the technological properties of

ades of pigmentary titanium dioxide can only be made
e process which creates a large amount of sulphuric
product. The rutile grades of titanium dioxide are
by the chloride process.

grades of titanium dioxide are produced using mica
(potassium aluminium silicate) as a template to form
let structure. The surface of the mica is coated with
le using a specialised patented process.

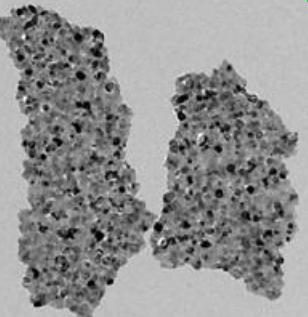
dioxide, platelet form is manufactured by subjecting
de (rutile) coated mica nacreous pigment to an
olution in acid followed by an extractive dissolution
of the mica is removed during this process and the
uct is a platelet form of rutile titanium dioxide.

Rutile



2.00 1/nm

1 500 nm



of 9 March 2012

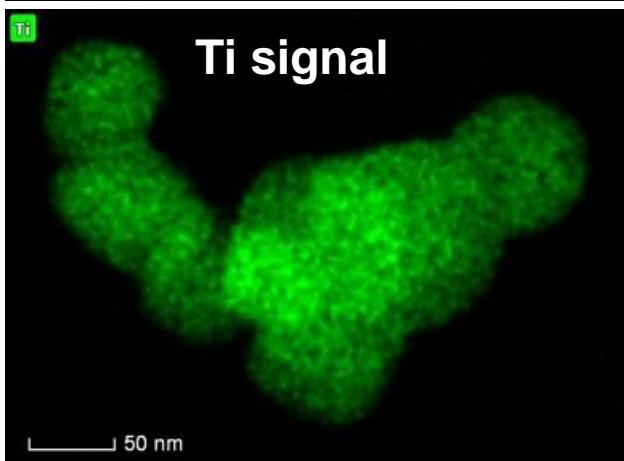
for food additives listed in Annexes II and III to Regulation (EC) 108 of the European Parliament and of the Council

(Text with EEA relevance)

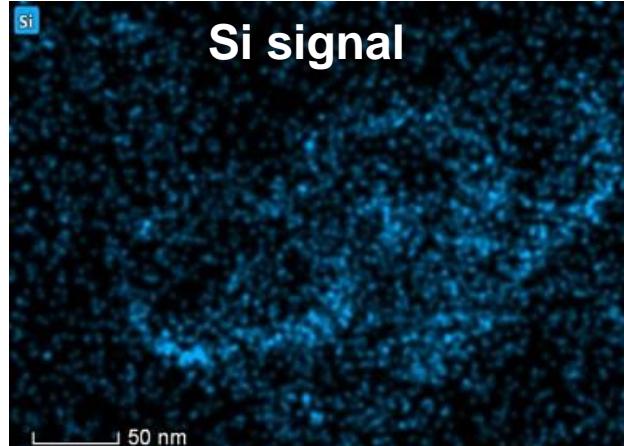
(OJ L 83, 22.3.2012, p. 1)

 50 nm

Ti signal

 50 nm

Si signal

 50 nm

CI Pigment White 6

Titanium dioxide consists essentially of pure anatase and/or rutile titanium dioxide which may be coated with small amounts of alumina and/or silica to improve the technological properties of the product.

HAADF Si Ti

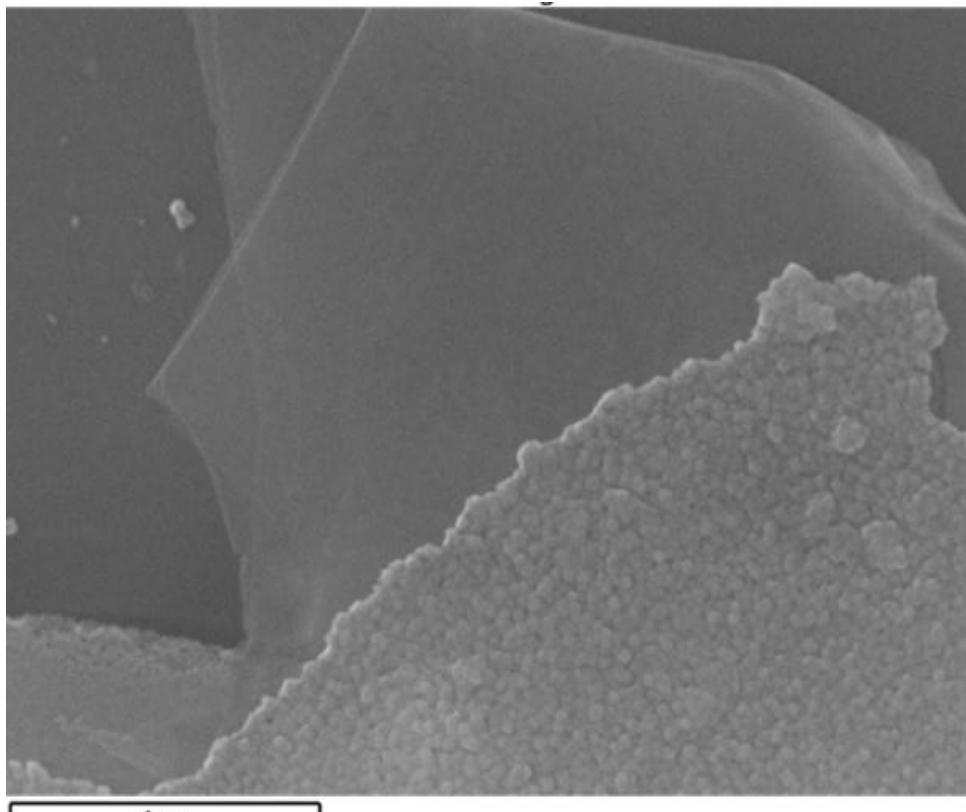
overlay

50 nm

be made
sulphuric
xide are

ng mica
to form
ted with

ubjecting
t to an
ssolution
and the
le.



1μm

Regulation (EU) No 231/2012

12

Annexes II and III to Regulation (EC) 231/2012 of the European Parliament and of the Council

evance)

2, p. 1)

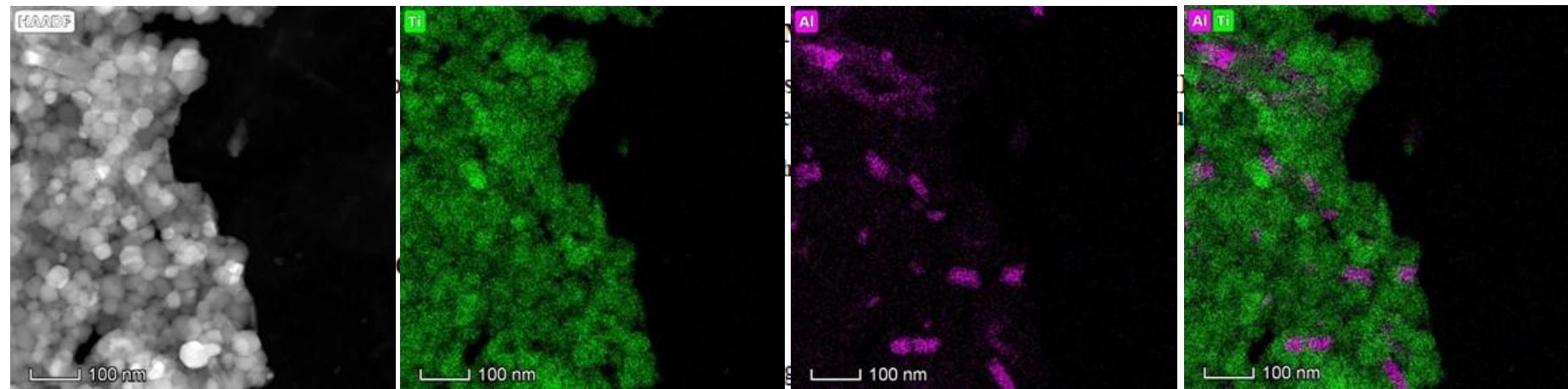
white 6

de consists essentially of pure anatase and/or rutile
de which may be coated with small amounts of
silica to improve the technological properties of

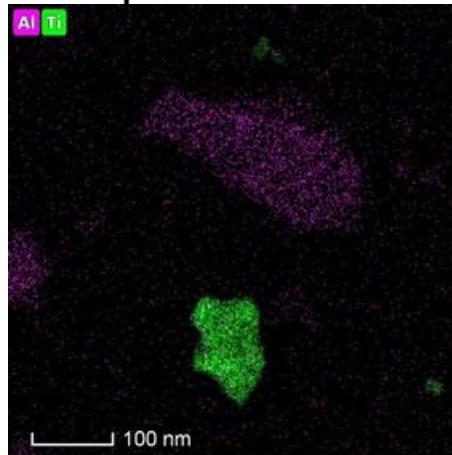
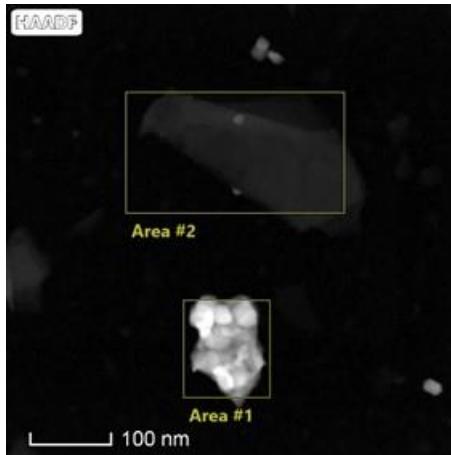
ides of pigmentary titanium dioxide can only be made
process which creates a large amount of sulphuric
product. The rutile grades of titanium dioxide are
typically made by the chloride process.

Certain rutile grades of titanium dioxide are produced using mica
(also known as potassium aluminium silicate) as a template to form
the basic platelet structure. The surface of the mica is coated with
titanium dioxide using a specialised patented process.

Rutile titanium dioxide, platelet form is manufactured by subjecting
titanium dioxide (rutile) coated mica nacreous pigment to an
extractive dissolution in acid followed by an extractive dissolution
in alkali. All of the mica is removed during this process and the
resulting product is a platelet form of rutile titanium dioxide.



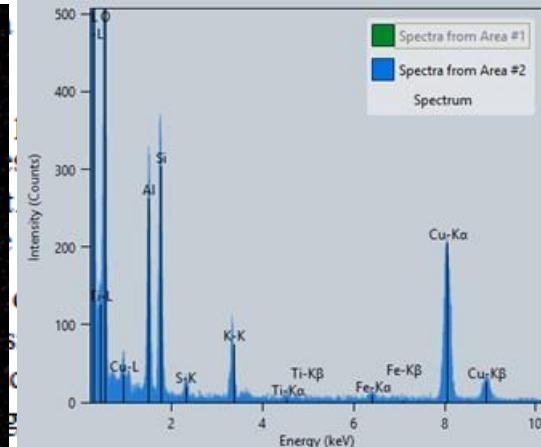
Definition



Titanium dioxide consists essentially of pure anatase and/or rutile titanium dioxide which may be coated with small amounts of

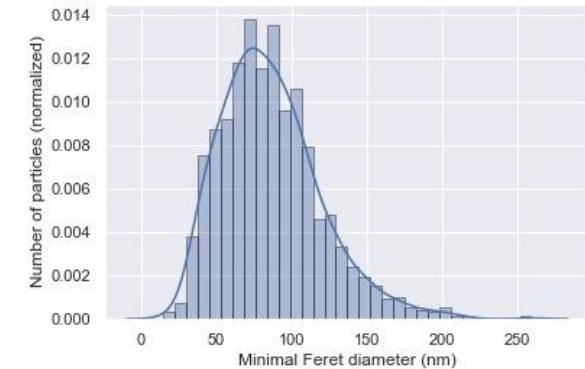
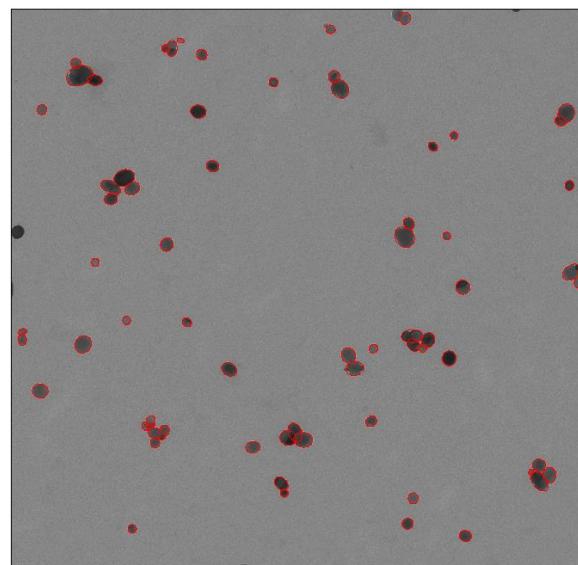
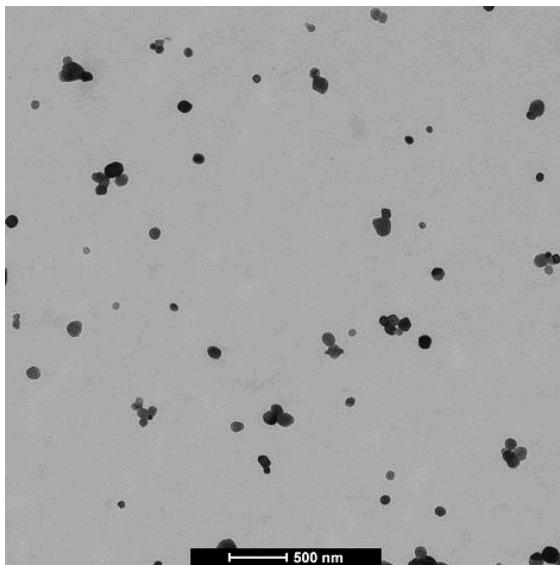
be made
sulphuric
oxide are

ing mica
e to form
ated with



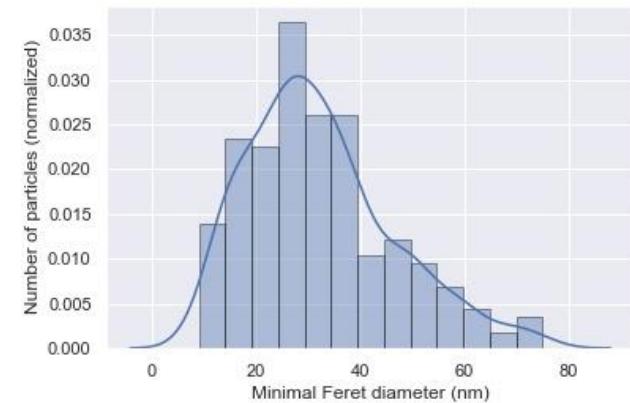
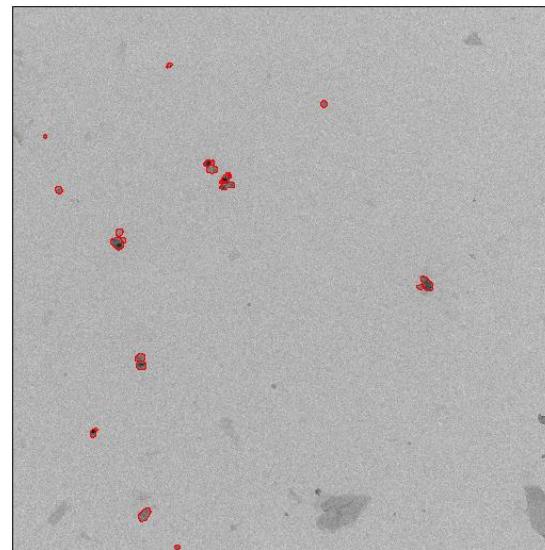
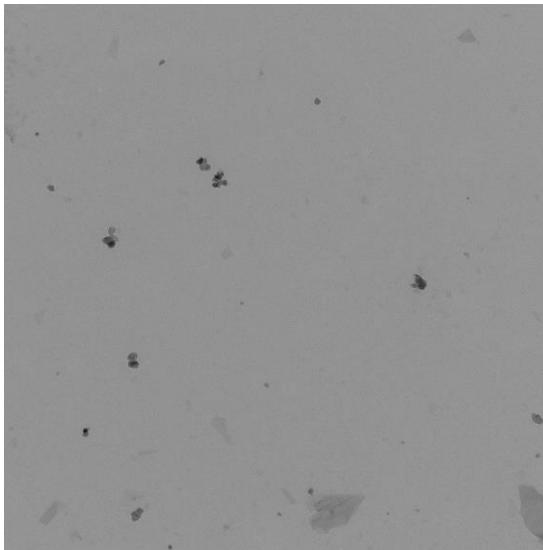
Rutile titanium dioxide, platelet form is manufactured by subjecting titanium dioxide (rutile) coated mica nacreous pigment to an extractive dissolution in acid followed by an extractive dissolution in alkali. All of the mica is removed during this process and the resulting product is a platelet form of rutile titanium dioxide.

E171 pristine: anatase



E171 Material:	1	2	3	4	5	6
Modal Min. Feret diameter (nm)	83	82	83	82	74	76
Median Min. Feret diameter (nm)	87	88	92	89	83	86

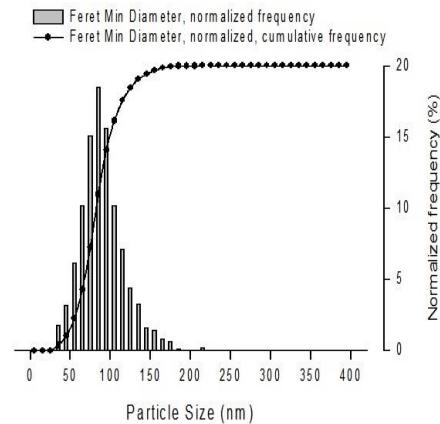
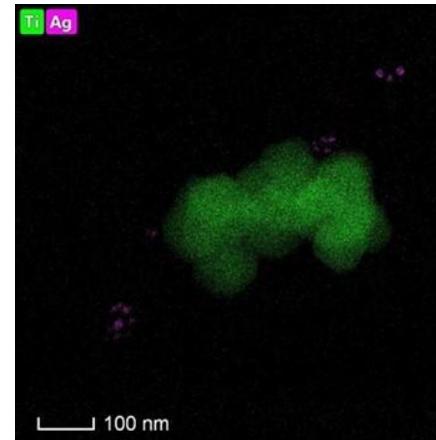
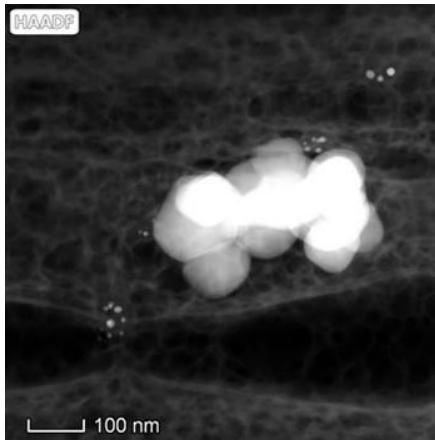
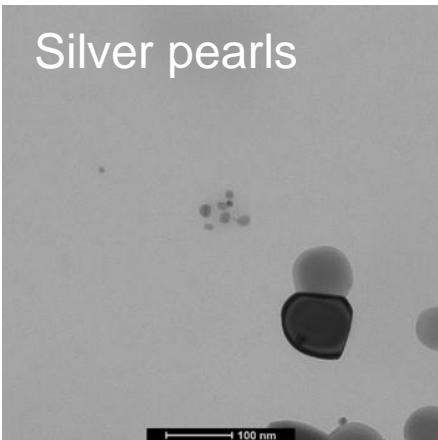
E171 pristine materials: rutile



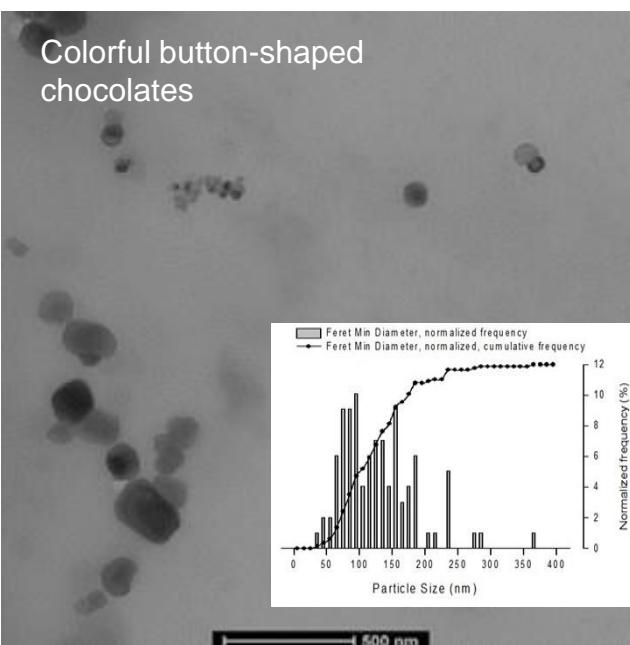
E171 Material:	7	8	9
Mode Min. Feret diameter (nm)	28	+/-30	+/-30
Median Min. Feret diameter (nm)	30	+/-30	+/-30

E171 in products: anatase

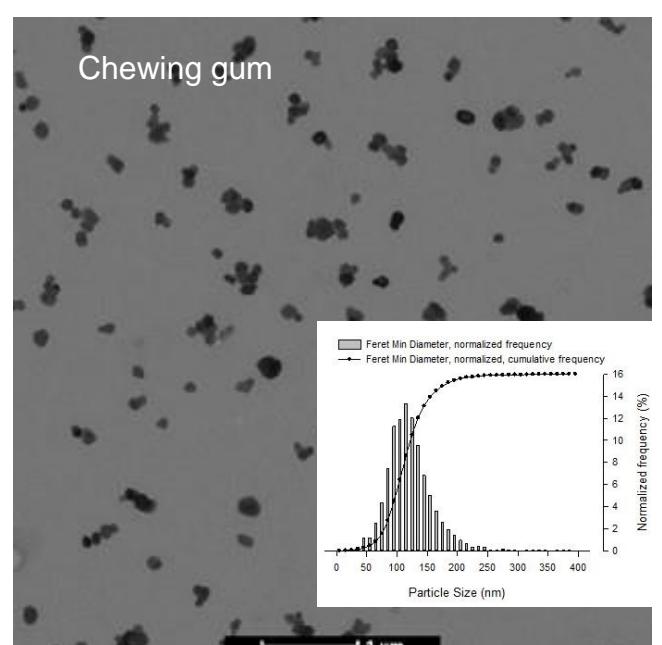
Silver pearls



Colorful button-shaped chocolates

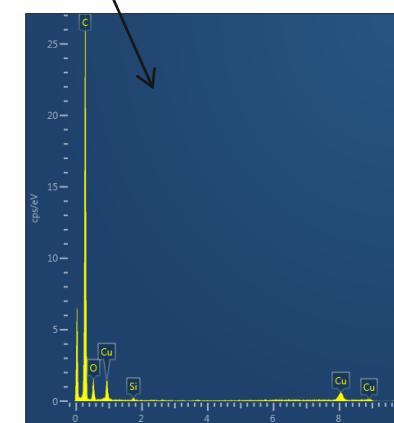
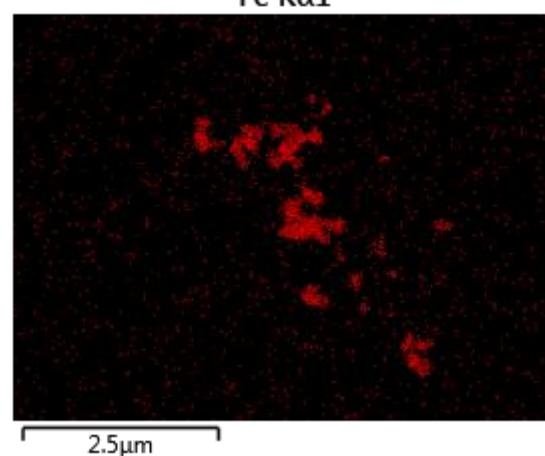
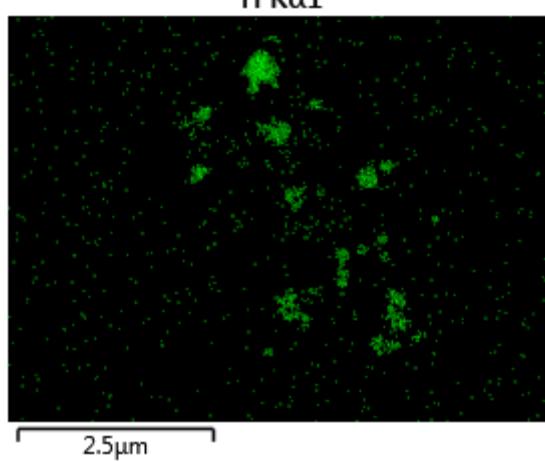
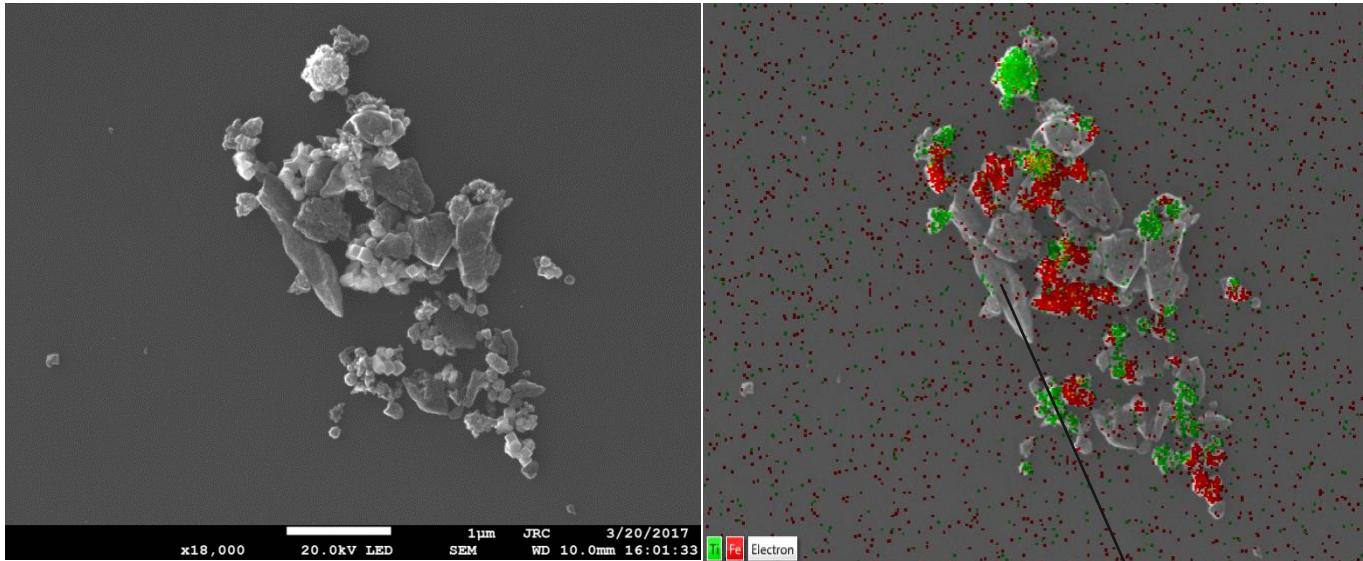


Chewing gum



E171 products: rutile

E171 in matrix E172 (iron oxide) and E153 (vegetable carbon)



Conclusions E171

- Several types of TiO_2 nano-sized particles are found in pristine E171 and in products containing E171
- E171: Less difficult sample preparation: no transformation between pristine materials and products

OUTLINE

- **Introduction**
- **Analytical methodology**
- **Analysis results of E174**
- **Analysis results of E171**
- **Lessons learnt**

Implementation of legislation and guidance

Challenge to build up specialized expertise and analytical capacity

→ Top-technologies

- Conventional TEM
- Advanced ICP-MS (Sciensano Tervuren)
- Analytical SEM (SEM-EDX, visiting scientist JRC Geel)
- Analytical TEM (Sciensano Ukkel, just installed, 1 mio euro)



→ Development and introduction of new methodologies (SOPs)

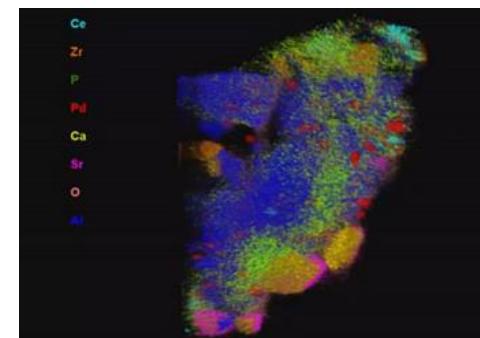
- Sample preparation
- Identification and analysis of individual (primary) particles: size, shape, surface and elemental composition

→ Complex validation approach, with few reference materials

→ Analytical capacity

- High degree of automation
- Standardized data analysis & reporting of number-based results

→ Search for collaborations, project opportunities, market



Next steps

Much more analyses...

Further methodology development:

- Automation to increase capacity
HAADF-STEM-EDX platform: Talos – Velox – Mapps - AVIZO
- Sample preparation

Market study (+ risk analysis)

Compulsory notification to Belgian FAFS

Difficulties applying/interpreting current legislation (definition)

Open communication

Experience in context of "EFSA Guidance on risk assessment of the application of nanoscience ...

Phys.-chem chapter in EFSA guidance provides general directions.

To be more practical, it could profit from:

- More detailed descriptions of approaches and methods,
- Guidance to «best practices» and later, to standardised methods (CEN)
- Illustrated in examples,
- Aim: to be implemented in MS (control and service) laboratories.

BE proposes WG in context of network for nanotechnologies

(possibly liaison with SANTE-JRC initiative « Nano in food: MS laboratories»)

? Support of other member states?

Acknowledgements

- FPS Health RCO RF16/6306/nanofood@ 2016-2018: "Implementation and validation of an approach to assess the fraction of engineered nanomaterials in food additives" (nanofood@).
- EFSA 2017-2020: Physico-chemical characterization and exposure analysis of nanomaterials in food additives in the context of risk assessment (EFSAfood@)
- FPS HealthDG4 2016-2018: "Physico-chemical characterization of the fraction of engineered nanomaterials in silver (E174) food additives in the context of risk assessment (nanoAg@)"
- BELSPO 2015-2016: "Method validation of nanomaterial analysis by single particle ICP-MS and DLS" (acronym: Nanoval)
- BELSPO BRAIN.BE (BR/154/A4/To2DeNano) 2016-2018: "Towards a toxicological relevant definition of nanomaterials" (To2DeNano)
- EC Large-scale integrating project FP7 NMP.2012.1.3-3 2013-2017: Regulatory testing of nanomaterials . "A common European approach to the regulatory testing of Nanomaterials" (NANOREG).
- CEN/TC 352: Nanotechnologies — Guidance on detection and identification of nano-objects in complex matrices
- EC Large-scale integrating project FP7 NMP.2013.1.4-3 2013-2017: "Development of methods and standards supporting the implementation of the Commission recommendation for a definition of nanomaterial. (NanoDefine).
- FAFS 2017-2019 NRL Chemistry 1 – parcel 4: Food contract materials – Heavy Metals – Nanoparticles



Contact

Jan Mast, Jan.Mast@sciensano.be

Service Trace Elements and Nanomaterials
Groeselenberg 99, 1180 Uccle
Leuvensesteenweg 17, 3080 Tervuren



Sciensano • Rue Juliette Wytsmanstraat 14 • 1050 Brussels • Belgium
T + 32 2 642 51 11 • T presse + 32 2 642 50 01 • info@sciensano.be • www.sciensano.be